

11. At which five amino acid residues narrow the channel to 1.99 Å on AQP0 polypeptide chain sequence **backbone**? Phe...., His...., Met...., Ala...., Arg.....

12. Four **backbone** carbonyls $>C=O<\cdots H-O-H$ of successive amino acid residues the canonical **AQP0** channel <u>Hydrogen bond acceptors</u> that align **water**s through the channel?

>C=O<••••H-O-H Gly.....,Ala....,Gly....,Met..... H-O-H••••>O=C< 13. Three amino acids water molecules ordered in line with three <u>donor Hydrogen bonds</u>

with four_acceptor bonds?

Cytosol

Cvtosol

4 **H₂O<••••H-N=C<** Asn.....,Arg.....,His...

14. Four amino acids <u>constricts the channel</u> diameter to 2.5 Å with side chain directly toward the central axis ? Phe....,Leu...,Tyr2....to central axis of the channel
15. Two amino acid side chains just after the Tyr23 constriction are NPA–NPA <u>signature</u> motifs that orient the key central water molecule what are responsible for the reorientation which so are necessary for prevent and restrict any proton conduction? Asn.....,Asn......

- 16. Four <u>cytoplasmic side</u> amino acids of the NPAs <u>signature motifs</u> line three of **backbone** carbonyl oxygens >C=O<••••H-O-H resumes along the wall point directly into the channel and it ends at Tyr149? Gly.....,Ala....,His....,Val......
- 17. Five amino acids form another constriction that is the channel narrowest region? It accepts close average radius of water pore sphere SF ~1.1 Å H–O–H longitudinal 1.4 Å and 0.55 Å bent size of dipole with a maximum diameter of 1.5 Å as water specific bent-angular structure? Tyr.....,Val.....,Gly....,His....,Phe......

18. What number of water molecules the line of eight backbone carbonyls >C=O---H-O-H and other channel-lining residues establishes a tight single-file pathway for water highlighted molecules in the Aquaporin 0 channel? ...water molecules in a tight single-file pathway... 19. What amino acid side chain residue from backbone chain of Aquaporin 0 is responsible for the inhibitory effect of **pH** or **Ca²⁺** on **water** transport rates? Therefore, Nemeth-Cahalan and Hall made mutations experiments that changed His-40 to alanine40, aspartate40, or lysine40, and they showed that treatment of oocytes expressing the mutants no longer displayed the pH-dependent closing as pH was raised from >6.5. Reaction of histidines40 in AQP0 with diethylpyrocarbonate (DEPC) removed **pH** dependence and actually increased water conductance (restored by histidine40-specific reversal with hydroxylamine), adding support to the case that titration of histidines40 alters conductance! His-..... increases conductance at pH > 20. Five 5 intermolecular forces are known in biochemistry and physiology of proteins? ... 3... 4.-**S-S-**.... 1.... 2... 5.... 21. What 3 intermolecular forces fold protein chains of AQP0? Call three 3 identified present!

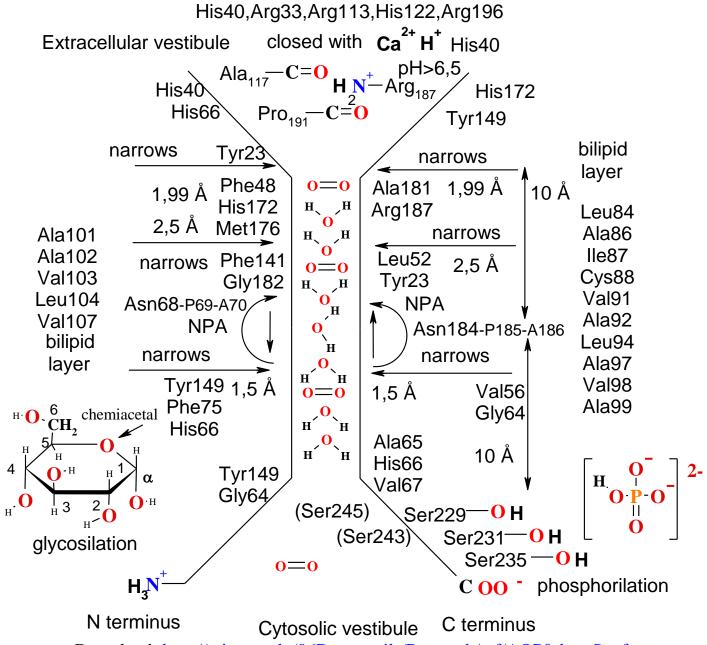
1.... 2...

3...

22. Put amino acids numbers in 28-Å-long, **cylindrical** AQP0 **channel** is <u>flanked by shallow</u> **vestibules** on each end! Channel volume show in the background, with major channel-forming residues amino acid names! The central region show with diameter of <2.5 Å, the regions both side shown with diameter of >2.5 Å and <10 Å long distance from central region center 0! Starting from the <u>extracellular</u> side, the vestibule funnel like diameter of <10 Å with 10 water molecules **HOH** 405, 415, 455, 411, 426, 407, 436, 439, 451, 435, 465 oxygen between residues show oriented into the center line. Drawn 7 **HOH**. Center amino acids are responsible for most of the narrowing of the vestibules up to channel narrowest diameter of 1.99 Å, 2,5 Å, 1.5 Å ! Seven charges **-NH**³⁺ and **-COO**[•] both side of membrane prevent conductance of ions!

hydrophobic trans cellular junction

Gly124,Pro123,Leu121,Phe198,Leu194 Leu39,Pro38,Gly37,Ala35,Gly114 Val112,Ala111,Pro110,Pro109;Val125 positive charged network AQP0 repulses junction AQP0



Download: http://aris.gusc.lv/06Daugavpils/Research/tgf/AQP0skemS.tgf

23. Call the <u>cell</u> and <u>organelles</u> localization sites where transport enzymes Aquaporins 0 found!

..... <u>cells as inter cellular</u> <u>or as</u>

between thin fiber..... cells.

24. Two Tyrosines and three Histidines are present down the monomer channel z axis?

Show the positions of Tyrosine residues on backbone trace and what

the positions of the Histidines that are close to the channel and vestibules!

Tyr....,Tyr....,His....,His....,

25. What three C-terminal **2B6P**.pdb amino acids are Posttranslational modificate and have also been suggested in the regulation of AQP0 activity, as C-terminal chain have been implicated as regulatory **phosphorylation** sites in human AQP0? What C-terminal mobile helix name and what three of 24 amino acids 239–263 suit for phosphorilation in structure

of **bAQP0**? **H**......Ser......(Ser245),Ser......(Ser243,Ser240),Ser......

26. In Eye-lens fiber cells 15 residues of amino acids non polar groups hydrophobic interaction forces between bound two aquaporins (AQP0's) create inter membranes cells 16 Å gap junction. Each contact point integrate energy -10 ^{kJ}/_{mol}, if 15 contact points than are -150 ^{kJ}/_{mol}.

Leu..., Pro..., Gly..., Ala..., Gly..., Val..., Ala..., Pro...,

Pro...., Val...., Gly...., Pro...., Leu....; Phe...., Leu.....

27. Four amino acids as conserved throughout all **AQPs**, alternate conformation located in the central "selectivity filter **SF**" of the channel to contribute channel closure?

with two good hydrogen bonds <u>from</u> Arg>NH...O=C<Pro.....

and from Asn>NH...O=C<Ala.....

28. Five amino acids on surface create positive +5 charge network repulsing AQP0

on **Eye-lens** extracellular side. His..., Arg...., Arg...., His...., Arg..... **29.** Arg-187, Asn68 3. Conformation in the central "selectivity filter **SF**" make channel closure with two <u>Hydrogen</u> bonds Arg-187>NH...**O**=C< un Asn68>NH...**O**=C< :

>NH...O=C<Pro...., O=C<Phe...., O=C<Asn, O=C<Arg...., O=C<Ser....;

>NH to O=C<Asn...., O=C<Asn...., O=C<Ala...., O=C<Asn.... and

What two amino acids which are conserved throughout all AQPs Val...., Tyr.....?

30. What **AQP0** isoelectric point IEP=pH=pK_{a-vid} at physiologic pH=7,36 ? To determine water solution pH with **AQP0** concentration C=0.00000063=10^{-6,2}M (^{mol}/_{Litre})!

21 L-α-Amino Acids proteins polypeptide protolysis pKa value isoelectric point IEP

At physiologic pH=7, 36 ±0.01 carboxylic groups **R-COO**⁻ negative charged and amino groups **R-NH**³⁺ positive charged. For example, glutamic acid pK_a reference to physiologic pH value smaller as $pK_{aR-COO}=4.25<7.36$, $pK_{aCOO}=2.19<7.36$ and for amine is greater as physiologic pH: 9.67= pK_{a-NH3} +>7.36.

Table sho	wn constants pKa	of four type	parallel protolytic equilibria in each amino acid molecule:
acid	⇔ base	+ H +;	Parallel protolytic equilibria number NpKa average isoelectric
1. R-COOH	⇔R-C <mark>OO</mark> -	+ H +;	point and constant pK_a value IEP= pK_a is calculated as
2. R-NH 3 ⁺	⇔ R-NH 2	$+H^{+};$	$IEP = pK_a = (\Sigma \ pK_{a R \ group} + pK_{a-NH3+} + pK_{a-COOH})/NpK_a$
3. Tyr-phenol-O	H⇔Tyr-phenol-	0 − + H ⁺ ,	In Ostwald's dilution law calculate pH of solution
4. Cys-SH	⇔Cys- <mark>S</mark> -	$+\mathbf{H}^+$	at concentration C logarithm: $pH = \frac{pK_a - \log C}{2} = \dots$
-	-		

Amino Acid	1		рК _{аNH3+} р	KaRgroup			
Isoleucine I		2.36	9.68		Biochemistry, University of Virginia 1995		
Valine V		2.32	9.62		<i>Myoglobin</i> IEP=7,36 is neutral zero ,,0" charged molecule,		
Leucine L		2.36	9.60		as IEP=7,36 is equal physiologic pH_{blood} =7,36 1MBO.pdb		
Phenylalanine	F	1.83	9.13				
	С	1.96	10.28	8.18	Albumin E7G.pdb 7,32=IEP 7 fatty acids small - charge and		
Methionine	Μ	2.28	9.21		7,40=IEP absent 7 fatty acids $+$ positive at pH=7.36, but		
	4	2.34	9.69		gamma Clabulin IgC1 ndb malagula has positiva (1) abarga		
	F	1.99	10.96		gamma <i>Globulin</i> IgG1.pdb molecule has positive (+) charge,		
2	G	2.34	9.60		as at physiologic pH=7.36 is greater IEP=7.91.		
	Т	2.11	9.62		Iso electric point IEP=pK _a as well protolytic constant pK _a calculates		
	S	2.21	9.15		one of side residues R constants sum $\Sigma p K_{aRside residue}$		
	W	2.38	9.39		plus pKaNterminusNH3+ and plus pKaCterminusCOO-		
2	Y	2.20	9.11	10.07	sum dividing with number NpKa of acidic groups in molecule		
Histidine	Η	1.82	9.17	6.00	$IEP=pK_a=(\Sigma pK_{aR side residue} + pK_{aNterminus} + pK_{aCterminus})/NpKa$		
I I	D	1.88	9.60	3.65			
Glutamate	Е	2.19	9.67	4.25	Figure Separation of serum proteins by electrophoresis.		
Asparagine	Ν	2.02	8.80		(a) A sample is applied as a narrow line at the origin. After		
Glutamine	Q	2.17	9.13		electrophoresis at pH 8.8, the paper is dried and stained.		
Lysine	Κ	2.18	8.95	10.53	(b) A plot of color intensity of each spot.		
Arginine	R	2.17	9.04	12.48	γ Globulin moves slower as Albumin		
·				Р	roteins move this direction \leftarrow spot line position sample origin at start		
Р	osi	tive (+)	electrode	2			
Positive (+) electrode Direction of movement							
	ice		lovement	15			
		<=-	1				
Albumin IEP=7,32÷7,40< <pH=8,8 γ Globulin IEP=7,91 < pH=8,8 γ (b)</p							
IEP=7,32 \div 7,40<< pH=8,8							
$\gamma Globulin IEP=7,91 < pH=8,8 \frac{5}{3}$ (b)							
Myoglobin IEP=7,36 $<$ pH=8,8 Albumin gamma Globulin							

Seleno cysteine, the 21st L-a-Amino Acid gamma *Globulin*

Seleno cysteine is an L- α -amino acid found in a handful of proteins, including certain **peroxidases** and **reductases** where it participates in the catalysis of electron transfer reactions. As its name implies, a selenium Se atom replaces the sulfur S of its structural analog, cysteine. The p*K*₃ of seleno cysteine 5.2 is 3 units lower than that of cysteine 8.18. Since seleno cysteine is inserted into polypeptides during translation, it is commonly referred to as the "21st amino acid." However, like the other 20 genetically encoded amino acids, seleno cysteine is specified by a simple three-letter codon UGA (see class 16 week Nucleo proteins tRNA 62 codons).

Lens-specific Aquaporin-0 (AQP0) functions as a specific water pore and forms the thin junctions between fibre cells http://aris.gusc.lv/ChemFiles/Aquaporins/1YMGpI.doc; http://aris.gusc.lv/ChemFiles/Aquaporins/1YMGpI.xls Sequence of 263 AA Amino Acids in AQPO molecule 1YMGpI.pdb: 80*3+23=263 10 20 30 40 50 60 70 80 MWELRSASFWRAICAEFFASLFYVFFGLGASLRWAPGPLHVLQVALAFGLALATLVQAVGHISGAHVNPAVTFAFLVGSQ MSLLRAICYMVAQLLGAVAGAAVLYSVTPPAVRGNLALNTLHPGVSVGQATIVEIFLTLQFVLCIFATYDERRNGRLGSV ALAVGFSLTLGHLFGMYYTGAGMNPARSFAPAILTRNFTNHWVYWVGPVIGAGLGSLLYDFLLFPRLKSVSERLSILKGS RPSESNGQPEVTGEPVELKTQAL

	Kac <mark>oo</mark> - pK		pK _{RR} ;Nc	hainAA		
M		9,21	4.95	1	1	In account are present 3 Cysteine residues $Cys = pK_{RR} = 8.18$;
E			4,25	3	2	Sum of AC r Ko volves in table
R			12,48	5	3	Sum of 46 pKa values in table
R C			12,48	11	4 5	O_{1}
E			8,18 4,25	14 16	6	Calculation tasks for Aquaporine 0 molecule AQP0
Y			4,23	23	7	
R			12,48	33	8	Protolytic constant pK _a isoelectric point IEP=pK _a calculate
Н			6	40	9	of side chains $\Sigma p K_{aRside group}$. $p K_{aNterminal} N H_3$ and $p K_{aCterminal} COO-constants$
Н			6	61	10	sum divide with number of acid groups NpK _a :
Н			6	66	11	sum divide with humber of deta groups reprize.
R			12,48	85	12	IEP= $pK_a = (\Sigma pKa_{Rside group} + pKa_{Nterminal} + pKa_{Cterminal})/NpKa$
С			8,18	88	13	
Y			10,07	89	14	30.1 Acid groups number in sum NpKa=44+2=
Y			10,07	105	15	263 amino acids of them protolytic constants pK_a for side groups 44+2.
R			12,48	113	16	N-terminal Methionine M pK _{aNterminal} =9.21 and
Н			6	122	17	
E			4,25	134	18	C-terminal Leucine L pKa _{Cterminal} =2.36
C			8,18	144	19	Sum are calculate as
Y			10,07	149	20	
D E			3,65 4,25	150 151	21 22	$\Sigma pKaRside group + pKaNterminal + pKaCterminala =$
R			4,25	151	22	20.2 Average and group constant $pV = -pV$ -IED ISOFI FUTDIC DOINT
R			12,48	152	23	30.2 Average acid group constant pK _{mean} =pK _a =IEP ISOELEKTRIC POINT
R			12,48	156	25	NpKa=44+2=
Н			6	172	26	
Y			10,07	177	27	IEP=389,69 / 46 =
Y			10,07	178	28	At pH value of amino acid and protein on isoelectric point pH=IEP
R			12,48	187	29	
R			12,48	196	30	total charge is zero " 0 " IEP =pH
Н			6	201	31	0
Y			10,07	204	32	
Y			10,07	219	33	<u>Underline</u> and determine existing: positive (+) or <u>zero charge</u> or negative (-)!
D			3,65	220	34 35	-COOH & -NH3 ⁺ positive -COO ⁻ & -NH3 ⁺ charge negative -COO ⁻ & -NH2
R K			12,48 10,53	226 228	35 36	$-COOII & -INII3 \cdot positive -COO & -INII3 \cdot Charge negative -COO & -INII2$
Б			4,25	228	30 37	30.3 AQP0 molecule charge sign (+). zero "0" or (-) at physiologic pH=7.36
R			12,48	232	38	
K			10,53	238	39	<u>Underline</u> existing:
R			12,48	241	40	-COOH,-NH ₃ ⁺ positive (+) pH=7.36 <iep=8,52 -coo<sup="" negative(-)="">-,-NH₂.</iep=8,52>
Е			4,25	244	41	20.4 AOD0 malacula abance (1) zero 0 , α (1) at algebra bance is all 9.9
Е			4,25	250	42	30.4 AQP0 molecule charge (+). zero " 0 " or (-) at electrophoresis pH 8.8
E			4,25	254	43	<u>Underline</u> existing:
Е			4,25	257	44	0
K	0.65		10,53	259	45	-COOH,-NH ₃ ⁺ positive (+) IEP = $8,52 < pH = 8,8$ negative (-) -COO ⁻ ,-NH ₂ .
L	2,63			263	46	

30.5 Calculate C=0.00000063=10^{-6,2} M AQP0 solution pH

by Ostwald dilution law concentration C in logarithm:

$$pH = \frac{pK_a - \log C}{2} = \frac{8,5228261 - \log 0.0000063}{2} = \frac{8,5228261 + 6,2}{2} = 14,723 / 2 = \dots$$

Attractor 7,36 AQP0 concentration isM.

Furthered follows aquaporin one AQP1 from erythrocytes